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Chapter 3: Chemistry 121. STUDY. PLAY. periodic law. the properties of the elements are periodic functions part of their atomic number. group. Vertical column in the periodic table. period. horizontal rows on periodic table. main group elements. elements in the s-block or p-block

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Calcium oxide reacts with water in a combination reaction to produce calcium hydroxide: CaO(s) + H₂O(l) → Ca(OH)₂(s) In a particular experiment, a 3.00-g sample of CaO is reacted with excess water and 3.46 g of Ca(OH)₂ is recovered. What is the percent yield in this exp

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Chapter 1 - Matter and Measurement: Part 1 of 3
AP Chemistry - Practice Test: Chapter 3 Name_____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1)Silver has two naturally occurring isotopes with the following isotopic masses: 107.47 Ar 107.47 Ar 106.90509 107.8682 amu.

Chapter 3 Matter and Energy
70 Chapter 3 Chemical Compounds. 3.1 Classification of Matter 71 Note that a chemical formula is a concise written description of the components of a chemical compound. It identifies the elements in the compound by their symbols and describes the relative number of atoms

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Chemistry 121 Chapter 3- Molecules, Compounds and Chemical Equations Elements and Compounds-o Elements combine together to make an almost limitless number of compounds o The properties of the compound are totally different from the constituent elements o H and O made of atoms held together by bonds o Chemical bonds are forces of attraction between atoms o The bonding attraction comes from attractions between protons and electrons Bond Types-o Two ...

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